

{4,4'-Dichloro-2,2'-[2,2-dimethylpropane-1,3-diylbis(nitrilomethanylylidene)]diphenolato- κ^4 O,N,N',O'}-nickel(II)

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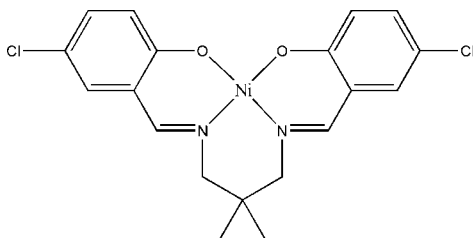
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Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.005$ Å; R factor = 0.039; wR factor = 0.081; data-to-parameter ratio = 14.2.

In the title compound, $[\text{Ni}(\text{C}_{19}\text{H}_{18}\text{Cl}_2\text{N}_2\text{O}_2)]$, the Ni^{II} atom shows a slightly distorted square-planar geometry. The dihedral angle between the mean planes of the coordination rings is 9.15 (12) $^\circ$ while the dihedral angle between the mean planes of the two aromatic rings is 3.48 (16) $^\circ$. In the crystal, pairs of intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds link neighboring molecules into a chain along the a axis. The crystal structure is further stabilized by $\pi-\pi$ interactions [centroid-centroid distance = 3.883 (2) Å].

Related literature

For standard bond lengths, see: Allen *et al.* (1987). For background to Schiff base metal complexes, see: Granovski *et al.* (1993); Blower (1998); Elmali *et al.* (2000). For related structures, see: Fun *et al.* (2008); Kargar *et al.* (2008); Rayati *et al.* (2011).



Experimental

Crystal data

$[\text{Ni}(\text{C}_{19}\text{H}_{18}\text{Cl}_2\text{N}_2\text{O}_2)]$
 $M_r = 435.96$
Monoclinic, $P2_1/n$
 $a = 6.9781$ (3) Å
 $b = 23.2517$ (11) Å
 $c = 11.8395$ (5) Å
 $\beta = 105.828$ (3) $^\circ$

$V = 1848.16$ (14) Å³
 $Z = 4$
Mo $K\alpha$ radiation
 $\mu = 1.36$ mm⁻¹
 $T = 296$ K
 $0.22 \times 0.15 \times 0.09$ mm

Data collection

Bruker SMART APEXII CCD
area-detector diffractometer
Absorption correction: multi-scan
(*SADABS*; Bruker, 2005)
 $T_{\text{min}} = 0.755$, $T_{\text{max}} = 0.888$

14131 measured reflections
3354 independent reflections
2373 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.062$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.039$
 $wR(F^2) = 0.081$
 $S = 1.02$
3354 reflections

237 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.31$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.29$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, $^\circ$).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C8}-\text{H8A}\cdots\text{O1}^{\text{i}}$	0.97	2.44	3.266 (4)	143
$\text{C12}-\text{H12A}\cdots\text{O2}^{\text{ii}}$	0.97	2.49	3.346 (4)	148

Symmetry codes: (i) $-x + 1, -y, -z + 2$; (ii) $-x + 2, -y, -z + 2$.

Data collection: *APEX2* (Bruker, 2005); cell refinement: *SAINT* (Bruker, 2005); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: KJ2181).

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
Blower, P. J. (1998). *Transition Met. Chem.* **23**, 109–112.
Bruker (2005). *APEX2*, *SAINT* and *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.
Elmali, A., Elerman, Y. & Svoboda, I. (2000). *Acta Cryst.* **C56**, 423–424.
Fun, H.-K., Kia, R. & Kargar, H. (2008). *Acta Cryst.* **E64**, o1895–o1896.
Granovski, A. D., Nivorozhkin, A. L. & Minkin, V. I. (1993). *Coord. Chem. Rev.* **126**, 1–69.
Kargar, H., Fun, H.-K. & Kia, R. (2008). *Acta Cryst.* **E64**, m1541–m1542.
Rayati, S., Ghaemi, A. & Notash, B. (2011). *Acta Cryst.* **E67**, m448.
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.